

Nuclear Radiation

<i>Name</i>	<i>Symbol</i>	<i>Composition</i>	<i>Mass</i>	<i>Charge</i>	<i>Speed*</i>	<i>Penetration</i>
alpha (α)	${}^4_2\alpha$ or ${}^4_2\text{He}$	He nucleus ($2p^+$ & $2n^0$)	4	2+	10% of c	low
beta (β)	β^- or ${}^0_{-1}e$	electron	0	1-	<90% of c	moderate
gamma (γ)	${}^0_0\gamma$ or γ	energy wave (photon)	0	0	c	very high
proton	${}^1_1\text{H}$ or ${}^1_1\text{p}$	proton	1	1+	10% of c	low-moderate
neutron	${}^1_0\text{n}$ or n	neutron	1	0	<10% of c	very high
positron [†]	β^+ or ${}^0_{+1}e$	positron	0	1+	<90% of c	moderate

* c = the speed of light, 2.998×10^8 m/s.

† The positron is a fast-moving “anti-particle,” mass equal to an electron (1/1836 amu), but opposite charge (1+).